

B.Tech	I	Semester	I/II	Teaching Scheme			Evaluation Scheme		
Subject Name	Engineering Chemistry – Lab			L	T	P	Credits	CCE	SEE
Subject Code	BTAS 21102			-	-	2	1	25	25
Type of course	BSC: Basic Science Course			CCE : Continuous and Comprehensive Evaluation SEE : Semester End Evaluation					
Prerequisite	Fundamentals of Organic / Inorganic / Physical / Analytical Chemistry								
Rationale	Material/Substance studies is the heart of understanding the Applied Chemistry which further helps in preparing ground to understand engineering better.								

Course Outcomes (COs): At the end of the course, students will be able to		Marks % weightage
CO – 1	Identify the quality of water and relate the different techniques of water treatment.	20
CO – 2	Determine the complexity of engineering problems and utilize correct metals and polymers	20
CO – 3	Describe the importance and application of engineering materials	40
CO – 4	Interpret the methods of functional chemistry as a logical means of problem solving	10
CO – 5	Distinguish the ranges of the electromagnetic spectrum used and apply it for exciting different molecular energy levels in various spectroscopic techniques and use it further to derive the logical issues	10

LIST OF PRACTICALS

1. Analysis of Steel Sample.
2. Analysis of Pyrolusite Ore.
3. Analysis of Brass Alloy.
4. Estimation of Total Hardness.
5. Determination of Concentration of Unknown Solution Spectrophotometrically.
6. Determination of pH and Conductance of the given solution.
7. Determination of Alkalinity of a given Water Sample.
8. Estimation of NH₃ and NH₄Cl in the given polluted water sample.

9. Determination of Saponification value of oil.
10. Determination of chloride content of water.
11. Study of Pyrolysis reaction of $ZnCO_3$.
12. Determination of moisture content in Coal.
13. Gravimetric estimation of Na_2CO_3 & $NaHCO_3$.
14. Study Wet Corrosion loss of Steel by weight loss method using electrochemical theory.
15. Stress Corrosion Cracking of Brass in NH_3 Solution.
16. Estimation of glucose by hypiodite method.
17. Organic/Inorganic Spotting.
18. Synthesis of Polymer

Recommended Reference Books

1. Laboratory Manual of Engineering Chemistry by S K. Bhasin & Sudha Rani, Dhanpat Rai Publications (P)Ltd., 2nd Edition, 2015
2. Engineering Chemistry with Laboratory Experiments by M S. Kaurav, PHI Learning Pvt. Ltd., 1st Edition, 2011
3. Vogel's textbook of Quantitative Chemical Analysis by Arthur I Vogel, Revised by Jeffery et. al., Addison Wesley, Longman Ltd. 5th Edition, 1989

CO-PO-Mapping

CO	PO 1	PO 2	PO 3	PO 4	PO 5	PO 6	PO 7	PO 8	PO 9	PO 10	PO 11	PO 12
CO-1	3	1	1	1	-	2	2	-	-	-	-	1
CO-2	3	3	1	-	-	1	2	-	-	-	-	1
CO-3	3	2	3	-	1	1	2	-	-	-	-	1
CO-4	3	2	2	1	2	1	1	1	-	-	-	1
CO-5	3	2	1	3	2	1	1	1	-	-	-	1

List of Open Source/learning website/Other Details if any:

- <https://vlab.amrita.edu/?sub=2>